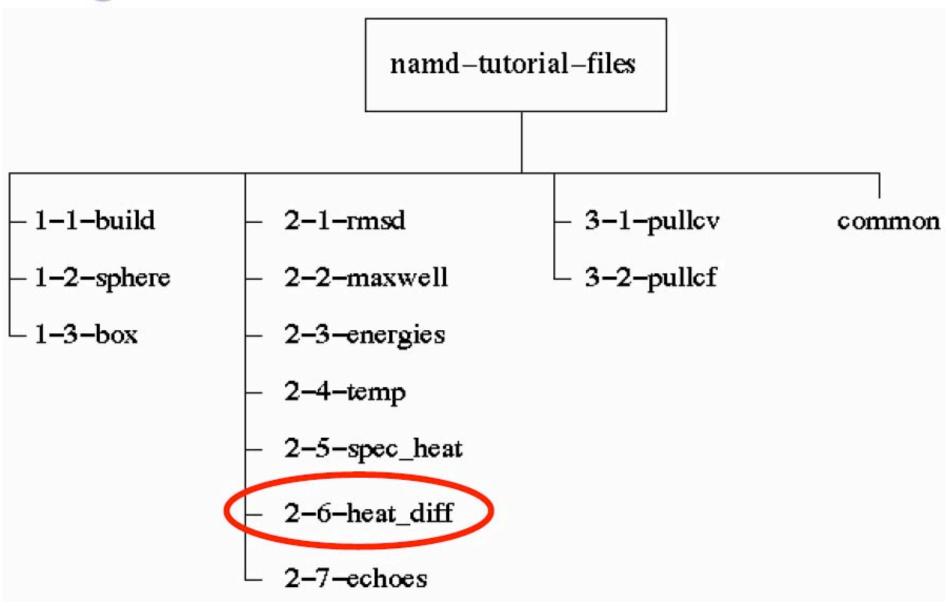
NAMD Tutorial (Part 2)

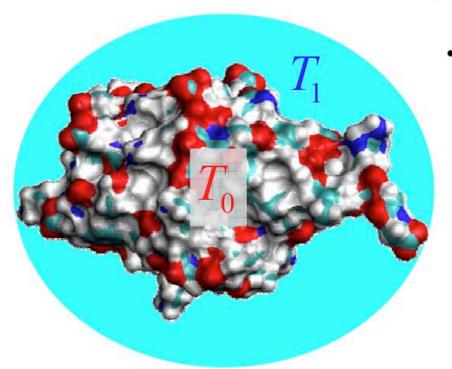
- 2 Analysis
 - 2.1 Equilibrium
 - > 2.1.1 RMSD for individual residues
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Organization of NAMD Tutorial Files



Simulated Cooling of Ubiquitin

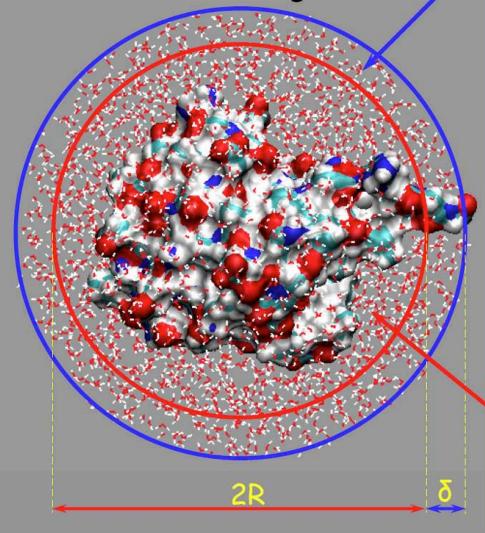
- Proteins function in a narrow (physiological) temperature range. What happens to them when the temperature of their surrounding changes significantly (temperature gradient)?
- Can the heating/cooling process of a protein be simulated by molecular dynamics? If yes, then how?



What can we learn from the simulated cooling/heating of a protein?

How to simulate cooling?

Heat transfer through mechanical coupling between atoms in the two regions



coolant layer of atoms

motion of atoms is subject to stochastic Langevin dynamics

$$m\ddot{\boldsymbol{r}} = \boldsymbol{F}_{FF} + \boldsymbol{F}_{H} + \boldsymbol{F}_{f} + \boldsymbol{F}_{L}$$

 $F_{\scriptscriptstyle FF}\! o\!{}$ force field

 $F_{\scriptscriptstyle H}$ \to harmonic restrain

 $F_f \rightarrow friction$

 $F_{\scriptscriptstyle L}$ \to Langevin force

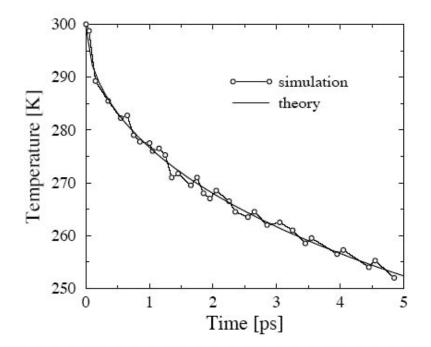
atoms in the inner region follow Newtonian dynamics

$$m\ddot{\mathbf{r}} = \mathbf{F}_{FF}$$

t	$\langle T_{sim} \rangle$						
0.05	298.75	1.05	276.00	1.95	267.00	3.25	261.00
0.15	289.25	1.15	276.50	2.05	268.50	3.45	258.50
0.35	285.50	1.25	275.25	2.25	266.50	3.55	259.50
0.55	282.25	1.35	271.00	2.35	264.50	3.95	256.50
0.65	282.75	1.45	271.75	2.55	263.50	4.05	257.25
0.75	279.00	1.65	269.50	2.65	264.50	4.45	254.00
0.85	277.75	1.75	271.00	2.85	262.00	4.55	255.25
1.00	277.50	1.85	268.00	3.05	262.50	4.85	252.00

Result from simulation

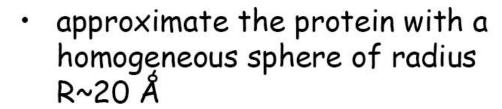
Table 1: Mean temperature $\langle T_{sim} \rangle$ [K] of the protein as a function of time t [ps].

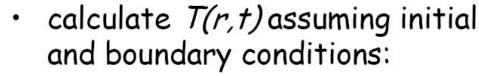


Heat Conduction Equation

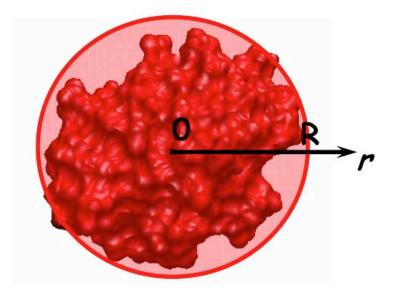
$$\frac{\partial T(\mathbf{r},t)}{\partial t} = D \nabla^2 T(\mathbf{r},t)$$
 mass density thermal diffusion coefficient
$$D = K/\rho c$$
 specific heat







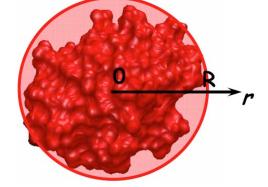
$$T(r,0) = T_0 \text{ for } r < R$$
$$T(R,t) = T_{bath}$$



$$\frac{\partial T(\mathbf{r},t)}{\partial t} = D \nabla^2 T(\mathbf{r},t) ,$$

$$D = K/\rho c ,$$

Initial condition



$$T(\mathbf{r},0) = \langle T_{sim} \rangle(0)$$
 for $r < R$,

Boundary condition

$$T(R,t) = T_{bath}$$
.

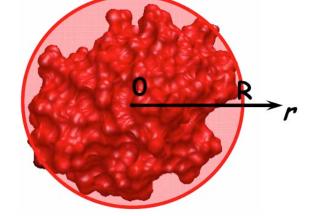
Spherical symmetry

$$\frac{\partial T(r,t)}{\partial t} = D \frac{1}{r} \partial_r^2 r T(r,t)$$

 T_{bath}

We assume

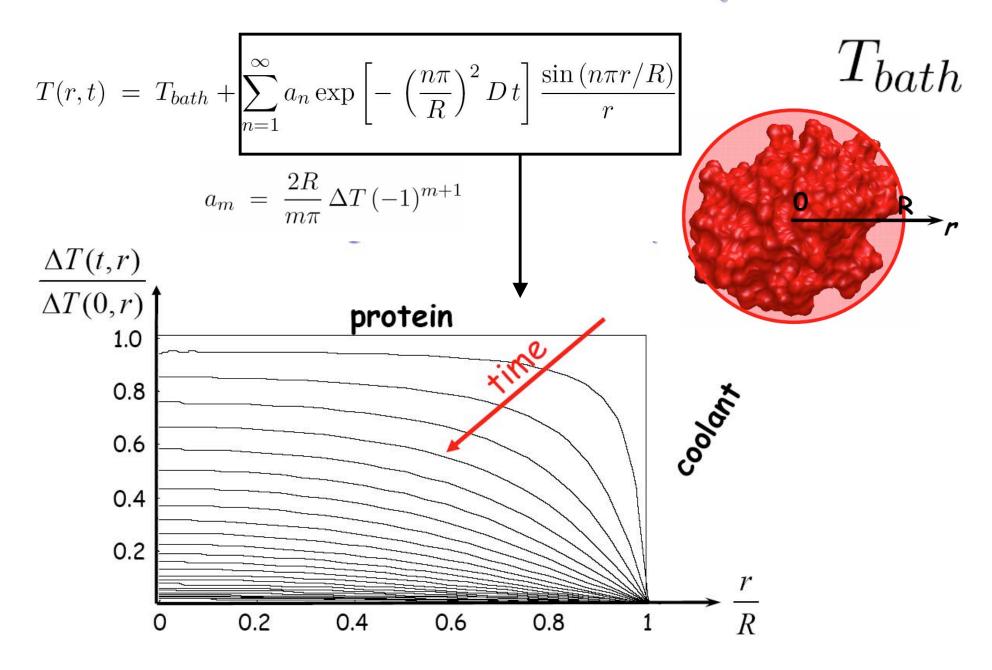
$$T(r,t) = T_{bath} + \sum_{n=1}^{\infty} a_n e^{\lambda_n t} u_n(r)$$
 difference from bath



Here u_n are the eigenfunctions of the spherical diffusion operator

$$L \equiv \frac{D}{r} \frac{d^2}{dr^2} r$$

$$\frac{D}{r}\frac{d^2}{dr^2}ru_n(r) = \lambda_n u_n(r) , u_n(0) = \text{finite}, u_n(R) = 0$$

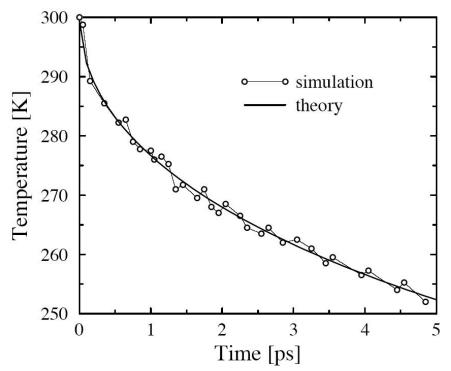


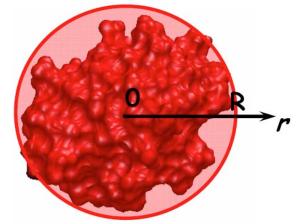
Temperature averaged over volume

$$\langle T \rangle (t) = \left(\frac{4\pi R^3}{3} \right)^{-1} \int d^3 \mathbf{r} \, T(\mathbf{r}, t) = \frac{3}{R^3} \int_0^R r^2 dr \, T(r, t)$$

$$= T_{bath} + \sum_{n=1}^{\infty} a_n \exp\left[-\left(\frac{n\pi}{R}\right)^2 D \, t \right] \frac{3}{R^3} \int_0^R r dr \sin\left(\frac{n\pi r}{R}\right)$$

$$= T_{bath} + 6 \frac{\Delta T}{\pi^2} \sum_{n=1}^{\infty} \frac{1}{n^2} \exp\left[-\left(\frac{n\pi}{R}\right)^2 D \, t \right]$$





 $D \approx 0.38 \times 10^{-3} \text{cm}^2 \text{s}^{-1}$

water $1.4 \times 10^{-3} \text{cm}^2 \text{s}^{-1}$

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