#### **Analysis of MD Trajectories**

#### Justin Gullingsrud

#### Beckman Institute University of Illinois at Urbana-Champaign



# Analysis of MD Data

- 1. Structural properties
- 2. Equilibrium properties
- 3. Non-equilibrium properties

#### Equilibrium properties can be studied via both **equilibrium** and/or **non-equilibrium** MD simulations



# Structural Properties of Biopolymers

- 1. End-to-end Distance
- 2. Radius of Gyration
- 3. Mean Square Displacement (MSD)
- 4. Root Mean Square Deviation (RMSD)
- 5. Debye-Waller Factor



#### 1. End-to End Distance



Represents the average distance *d* between the first and last segment of a (bio)polymer

Suitable to describe linear polymers



#### 2. Radius of Gyration

$$R_{G} = \sqrt{\sum_{a=1}^{N} m_{a} (r_{a} - r_{COM})^{2} / \sum_{a=1}^{N} m_{a}}$$

Mass weighted RMS average distance of the selected atoms from their center of mass (COM)

Suitable to describe branched chains with large number of ends



#### 3. Mean Square Displacement

$$MSD = \left\langle \sum_{a} (\mathbf{r}_{a} - \mathbf{r}_{a0})^{2} \right\rangle$$

Describes the "distance" between two conformations of a (bio)polymer (or group of selected atoms)

First the two conformations must be aligned



#### 4. Root Mean Square Deviation

$$RMSD = \left[ \left\langle \sum_{a} (\mathbf{r}_{a} - \mathbf{r}_{a0})^{2} \right\rangle \right]^{\frac{1}{2}}$$

Describes the "distance" between two conformations of a (bio)polymer (or group of selected atoms)

First the two conformations must be aligned



#### 5. Debye-Waller Factor

(B- or Temperature Factor)

$$B_a = \frac{8}{3} \pi^2 \langle \delta \mathbf{r}_a^2 \rangle, \quad \delta \mathbf{r}_a = \mathbf{r}_a - \langle \mathbf{r}_a \rangle_t$$

Describes the reduction of the intensity of Bragg scattering due to motion of atoms about their equilibrium position

Atomic scattering factor:

$$f = f_0 \cdot \exp[-B(\sin\phi/\lambda)^2]$$

Does not vanish even at T=0 because of the zero point motion of the atoms !





Beckman Institute, UIUC

#### **Non-equilibrium Properties**

- 1. Transport properties
- 2. Spectral properties

Can be obtained from *equilibrium* MD simulations by employing *linear response theory* 



#### Linear Response Theory





**Time Correlation Functions**  

$$C_{AB}(t-t') = \langle A(t) B(t') \rangle = \langle A(t-t') B(0) \rangle$$
since  $\rho_{eq}$  is t independent !  

$$A \neq B \quad \text{cross-} \\ A = B \quad \text{auto-} \} \text{ correlation function}$$
Correlation time:  $\tau_c = \int_{0}^{\infty} dt C_{AA}(t) / C_{AA}(0)$ 

Estimates how long the "memory" of the system lasts

In many cases (but not always):  $C(t) = C(0) \exp(-t/\tau_c)$ 



#### **Response Function**

or generalized susceptibility

External perturbation:  $V_{ext}(t) = -A \cdot f_{ext}(t)$ Response of the system:  $\langle A(t) \rangle = \int_{0}^{t} dt' R(t-t') f_{ext}(t')$ Response function:  $R(t) = \langle \{A(t), A\}_{PB} \rangle = -\beta \langle \partial_{t} A(t) A \rangle$ with  $\beta = 1/k_{B}T$ Generalized susceptibility:  $\chi(\omega) \equiv R(\omega) = \int_{0}^{\infty} dt e^{i\omega t} R(t)$ 

Rate of energy dissipation/absorption:

$$Q_{\omega} \equiv \langle A(t) \rangle \frac{df}{dt} = \frac{1}{2} \omega \chi''(\omega) |f_0|^2, \ f(t) = \operatorname{Re} f_0 e^{-i\omega t}$$



#### **Fluctuation-Dissipation Theorem**

Relates R(t) and C(t), namely:

$$\chi''(\omega) = (\beta \omega / 2) C(\omega)$$

In the static limit  $(t \to \infty)$ :  $C(0) = \langle A^2 \rangle = k_B T R(0)$ 

<u>Note</u>: quantum corrections are important when  $k_B T \le \eta \omega$  $\chi''(\omega) = \eta^{-1} \tanh(\beta \eta \omega/2) C(\omega)$ 



# Example: Absorption of Radiation by Electric Dipoles

Perturbation:  $V_{ext}(t) = -\mathbf{P} \cdot \mathbf{E}(t), \quad E(t) = E_0 \,\hat{\mathbf{e}} \cos \omega t$ 

Correlation function:  $C(t) = 1/3 \langle P(t) P \rangle$ 

Absorption coefficient:  $\alpha(\omega) = (4\pi\omega/c)[\chi''(\omega)/\epsilon'(\omega)]$ Applying the FDT:

$$\alpha(\omega) = (2\pi\omega^2\beta/c)[C(\omega)/\varepsilon'(\omega)]$$

P(t), and C(t) can be computed from a suitable MD trajectory



#### **Diffusion Coefficient**

γ

 $\mathbf{n}$ 

Generic transport coefficient:

$$= \int_{0}^{\infty} dt \langle \partial_{t} A(t) \partial_{t} A(0) \rangle$$

Einstein relation:  $2\gamma t = \langle [A(t) - A(0)]^2 \rangle$ 





NIH Resource for Biomolecular Modeling and Bioinformatics http://www.ks.uiuc.edu/

#### Analyzing Ion Channels

- The program HOLE can be used to obtain an estimate of pore radii along an axial coordinate.
- Other properties: diffusion coefficient, DELPHI analysis...





#### Analysis of water in channels

- Analysis of single water molecules demonstrates singlefile motion.
- Correlation functions can be obtained directly from the trajectories.





NIH Resource for Biomolecular Modeling and Bioinformatics http://www.ks.uiuc.edu/

#### Combining HOLE with VMD

Data from HOLE can easily be imported into VMD for structural analysis.





#### Secondary Structure Analysis

• Secondary structure trajectories can be displayed along with coordinate data.





NIH Resource for Biomolecular Modeling and Bioinformatics http://www.ks.uiuc.edu/

#### Trajectory Analysis in Practice

- Trajectory files are often very large, requiring special tools to work with them.
- Quite a bit of trajectory analysis can be done within VMD.
  - Use fast atom selections
  - Take advantage of fast methods (think vector code)
- Environments like Matlab are also well suited for trajectory analysis.



#### Collection of MD Data

- DCD trajectory file
  - coordinates for each atom
  - velocities for each atom

- Output files
  - global energies
  - temperature, pressure, …
  - unit cell dimensions



#### CatDCD

- Merge or split DCD files from NAMD: catdcd -o simA.dcd simA-01.dcd simA-02.dcd simA-03.dcd
- Create a new DCD file containing only selected atoms
  - Saves memory

- Makes atom selections go faster
catdcd -I protein.ind -o protein.dcd
simA.dcd



#### Other uses for CatDCD

- Count how many frames are in a DCD file: catdcd -num min\_all.dcd catdcd -num \*.dcd
- Grab the last 5 frames out of a DCD file: catdcd -first 196 min\_all.dcd



#### Analysis Features of VMD

- RMSD analysis, RMS best fits, mass-weighted RMSD, etc. can all be done easily in VMD.
- Phi/psi angles are available for a given atom selection:

set ca [atomselect top "name CA"]
set philist [\$ca get phi]

 Bond, angle and dihedrals values can be quickly determined for an entire trajectory:
 label add Bonds 0/10 0/20
 set bondval [label graph Bonds 0]



#### Matrices and vectors in Tcl

- In VMD, vectors are Tcl lists and matrices are nested lists.
- Vecadd, vecsub, transvec, etc. are mostly implemented in C and are reasonably fast.



#### VMD/Tcl Performance Tips

- Use a few large vectors (like x, y, z), rather than separate vectors for each atom, for your analysis scripts.
- Use CatDCD to whittle down a DCD file to just what you need.
- Atom selections are fast and scale well, but use them sparingly.



#### VMD and Python

- Many scientific packages for Python exist which can be useful for MD analysis
  - Numeric Python
  - Scientific Python
  - MMTK
- You can use these packages from within VMD by switching to the built-in Python interpreter (gopython).



#### MatDCD

- NAMD DCD files can be loaded to/from Matlab for analysis.
- Easy to compute correlation functions, perform principal component analysis, SVD, etc.
- No atom selection language, though VMD could be used to generate the Matlab script...



# Mindy

- "Minimal Molecular Dynamics" program based on NAMD source code.
- Provides a framework for "hacking" one's own analysis tools without having to understand all of NAMD.
- Has been used to find hydrogen bonds, compute interaction energies between subsets of atoms.

